



KSIT
KAMMAVARI SANGHAM (R) - 1952

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K.S. INSTITUTE OF TECHNOLOGY

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#14, Raghuvanahalli, Kanakapura Road, Bengaluru - 560109

04

BLUE BOOK

Name of the student: Adithya.M (Roll No 4)

Sem / Section: I/A' Branch: CSE

USN:

1	K	S	2	2	C	S	0	0	4
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SUBJECT: Physics

SUBJECT CODE: 22PHYS12

MAXIMUM MARKS

Test	I	II	III	Average Marks Obtained
Date	18/01/2023	01/03/2023	27/03/2023	(a) Test (60) 50
Marks Obtained	18	20	12	(b) Assignment (20) 20
Signature of Student	<u>Adithya</u>	<u>Adithya</u>	<u>Adithya</u>	(c) Activity (lab) (20) 19
Initials of Faculty	<u>Mb</u>	<u>Madhavi</u>	<u>Mb</u>	Total (60+20) (30) 70 27
Final Assessment				27+19 = 46

NAME OF FACULTY: Dr. Madhavi S

SIGNATURE OF FACULTY Madhavi

SIGNATURE OF H.O.D. Dasgi

K.S. INSTITUTE OF TECHNOLOGY

First Internal Test

Q.No	Marks	OR	Q.No	Marks	CO	CO	Total
1 (a)	04		OR	2 (a)	—	00	001
1 (b)	04	2 (b)		—	001		
1 (c)	03	2 (c)		—	001	002	07
3 (a)	03	OR	4 (a)	—	002		
3 (b)	04		4 (b)	—	002		
Grand Total							15

Second Internal Test

Q.No	Marks	OR	Q.No	Marks	CO	CO	Total
1 (a)	-04-		OR	2 (a)	—	003	003
1 (b)	-04-	2 (b)		—	003		
1 (c)	-04-	2 (c)		—	003	002 004	08
3 (a)	-04-	OR	4 (a)	—	002		
3 (b)	-04-		4 (b)	—	004		
Grand Total							20

Third Internal Test

Q.No	Marks	OR	Q.No	Marks	CO	CO	Total
1 (a)	-03-		OR	2 (a)	—		005
1 (b)	-01-	2 (b)		—			
1 (c)	-00-	2 (c)		—		004	08
3 (a)	-04-	OR	4 (a)	—			
3 (b)	-04-		4 (b)	—			
Grand Total							12

Assignment	A1 (10)		A2 (10)		Total
		10		10	10
Activity	15	05	20		Total
lab	15	04	19		19

Madhava
SIGNATURE OF FACULTY.

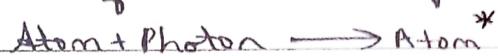
Final Interview

Part - A

1a) LASER stands for Light Amplification by Stimulated emission of Radiation. It's basic principle is interaction with matter. It has 3 principles.

- Induced absorption: It is the absorption of energy by the lower energy atom levels to the excited state.
- Spontaneous emission: It is the spontaneous release of energy by the higher excited state to the lower ground state.
- Stimulated emission: It is the release of energy when an excited atom is supplied with energy to yield 2 photons releasing the atom back to the ground state.

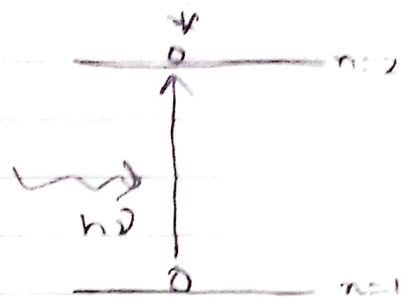
Case of Induced absorption



~~Rate~~

$$\text{Rate of Induced absorption} \propto N_1 N \nu_{12}$$

$$= B_{12} N \nu_{12} \quad \text{--- (1)}$$

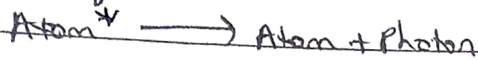


N_1 \rightarrow no. of atoms in lower energy state

ν_{12} \rightarrow energy density

B_{12} \rightarrow constant (transition from $n=1$ to $n=2$)

Case of Spontaneous emission:

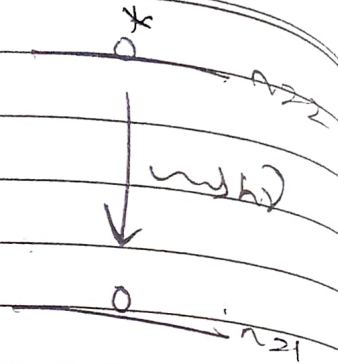


Rate of Spontaneous emission of N_2

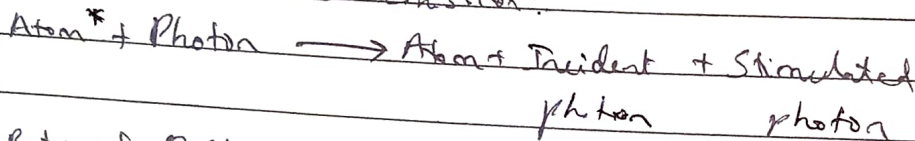
$$= A_{21} N_2 \quad \text{--- (2)}$$

$N_2 \rightarrow$ No. of atoms in the higher energy state

$A_{21} \rightarrow$ constant (Transition from $n=2$ to $n=1$)



Case of Stimulated Emission:



Rate of Stimulated emission of $N_2 U_{21}$

$$= B_{21} N_2 U_{21} \quad \text{--- (3)}$$

$U_{21} \rightarrow$ energy density

$N_2 \rightarrow$ no. of atoms in the higher excited state

$B_{21} \rightarrow$ constant (Transition from $n=2$ to $n=1$)

At Thermal equilibrium,

Rate of induced absorption = Rate of Spontaneous emission + Rate of Stimulated emission

From (2), (2), (3)

$$B_{12} N_1 U_{21} = A_{21} N_2 + B_{21} N_2 U_{21}$$

$$B_{12} N_1 U_{21} - B_{21} N_2 U_{21} = A_{21} N_2$$

$$U_{21} (B_{12} N_1 - B_{21} N_2) = A_{21} N_2$$

$$U_D = \frac{A_{21} N_2}{(B_{12} N_1 - B_{21} N_2)}$$

$$U_D = \frac{A_{21} N_1}{B_{21} N_1 \left(\frac{B_{12} N_1}{B_{21} N_2} - 1 \right)} \rightarrow (4)$$

From Boltzmann's law, we have

$$\frac{N_2}{N_1} = e^{-h\nu/kT} \quad \frac{N_1}{N_2} = e^{h\nu/kT} \rightarrow (5)$$

So, (5) in (4),

$$U_D = \frac{A_{21}}{B_{21} \left(\frac{B_{12} e^{h\nu/kT}}{B_{21}} - 1 \right)} \rightarrow (6)$$

From Planck's law, we have

~~$$U_D = \frac{8\pi h\nu^3}{c^3} \left[\frac{1}{e^{h\nu/kT} - 1} \right] \rightarrow (7)$$~~

$$U_D = \frac{8\pi h\nu^3}{c^3} \left[\frac{1}{e^{h\nu/kT} - 1} \right] \rightarrow (7)$$

Comparing (7) & (6), $\frac{A_{21}}{B_{21}} = \frac{8\pi h\nu^3}{c^3}$

Comparing (A) & (B), $\frac{A_{12}}{B_{21}} = 1$

$$B_{12} = B_{21}$$

$$B = B$$

(No need of subscript)

$$A_{21} = A$$

Rewriting (A),

$$U_{\nu} = \frac{A}{B \left(\frac{B}{B} e^{h\nu/kt} - 1 \right)}$$

$$U_{\nu} = \frac{A}{B \left(e^{h\nu/kt} - 1 \right)}$$

→ This is the required equation

A, B → Einstein's Coefficients

U_{ν} → Energy density

h → Planck's constant

ν → frequency

k → Boltzmann constant = 1.38×10^{-23}

t → Temperature

1 b) $T = 300\text{K}$ | $\lambda = 1 \times 10^{-6}\text{m}$ | $h = 6.63 \times 10^{-34}\text{Js}^{-1}$ | $k = 1.38 \times 10^{-23}$ / $^{\circ}\text{C}$
 $c = 3 \times 10^8\text{ms}^{-1}$

↑ given data

$$\frac{N_1}{N_2} = e^{-h\nu/kt} \quad (\text{BOLTZMANN LAW})$$

$$\frac{N_2}{N_1} = e^{-\frac{hc}{\lambda kT}}$$

$$\frac{N_2}{N_1} = e^{\frac{-6.6 \times 10^{-34} \times 3 \times 10^8}{1 \times 10^{-6} \times 1.38 \times 10^{-23} \times 300}}$$

to the power of

$$\frac{N_2}{N_1} = e^{\left(\frac{-19.99 \times 10^{-26}}{414 \times 10^{-29}} \right)}$$

$$\frac{N_2}{N_1} = e^{\left(\frac{10^3 \times -19.99}{414} \right)}$$

$$29 - 26$$

$$= 3$$

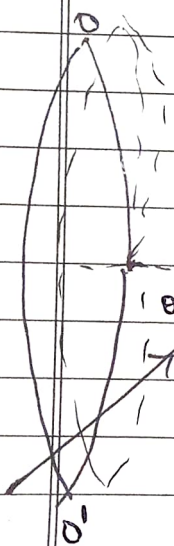
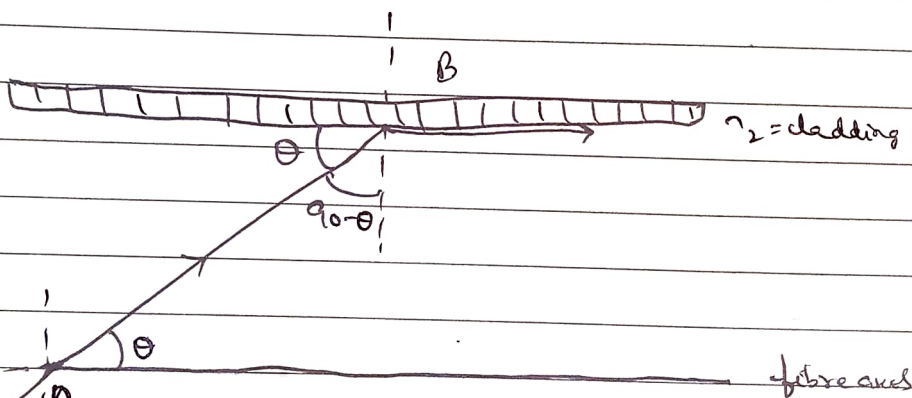
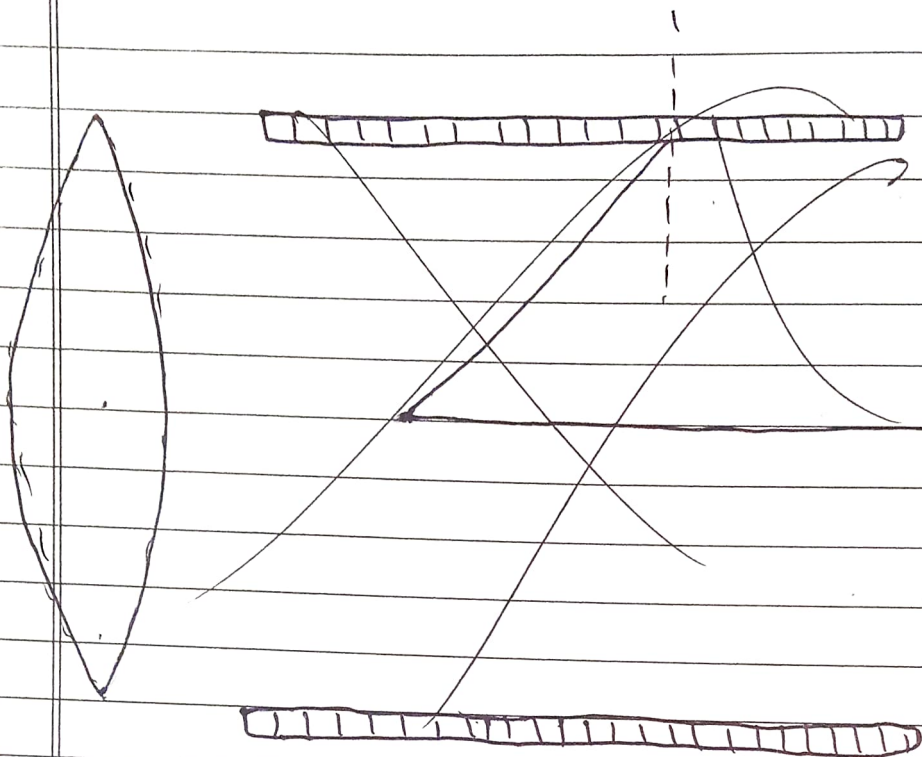
04

$$\frac{N_2}{N_1} = e^{(-0.64804 \times 10^3)}$$

$$\frac{N_2}{N_1} = \frac{1.3692826 \times 10^{-21}}{1.42516403}$$

$$\frac{N_2}{N_1} = 1.425 \times 10^{-21}$$

1) c)



$n_0 = \text{air surrounding medium}$

Acceptance
cone
D.A.C.

n_1 = refractive index of core
 n_2 = refractive index of cladding
 n_0 = refractive index of surrounding medium

$(n_1 > n_2 \rightarrow \text{always})$

Apply Snell's Law at A,

$$n_0 \sin \theta_1 = n_1 \sin \theta \rightarrow (1)$$

Apply Snell's Law at B,

$$n_1 \sin(90 - \theta) = n_2 \sin 90$$

$$n_1 \cos \theta = n_2 \quad \rightarrow (2)$$

$$\cos \theta = \sqrt{1 - \sin^2 \theta}$$

$$\sin^2 \theta + \cos^2 \theta = 1$$

$$n_1 \cos \theta = n_2$$

$$\cos \theta = \frac{n_2}{n_1}$$

03

$$\cos \theta = \frac{n_2}{n_1}$$

$$\sin \theta = \sqrt{1 - \frac{n_2^2}{n_1^2}}$$

If surrounding medium is air ($n_0 = 1$)

From (1),

Condition for wave propagation,

$$\theta < \sin^{-1} \sqrt{n_1^2 - n_2^2} / n_0$$

$$\cos \theta = \frac{n_2}{n_1}$$

$$\frac{\sin \theta_1}{\sin \theta} = \frac{n_1}{n_0}$$

$$\sqrt{1 - \sin^2 \theta} = \frac{n_2}{n_1}$$

$$\frac{\sin \theta_1}{\sin \theta} = \frac{n_1}{n_0}$$

$$\sqrt{1 - n_0^2 \sin^2 \theta_1} = \frac{n_2 \sin \theta_1}{n_1 \sin \theta}$$

on substituting,

$$\cos \theta = \frac{n_2}{n_1}$$

$$\sin \theta = \sqrt{1 - \frac{n_2^2}{n_1^2}}$$

$$\text{or } \theta = \sin^{-1} \sqrt{1 - \frac{n_2^2}{n_1^2}}$$

↑

Part B

3a) De Broglie related that light shows dual nature

$$\lambda = \frac{h}{mc}$$

The wavelength of matter waves is inversely proportional to the momentum of the particle.

$$E = h\nu \text{ (Planck's law) } \quad E \rightarrow \text{Energy}$$

$$h \rightarrow \text{Planck's constant} = 6.6 \times 10^{-34} \text{ J}\cdot\text{s}$$

$$\nu \rightarrow \text{frequency}$$

$$E = mc^2 \text{ (Einstein's energy-mass equation)}$$

 $E \rightarrow \text{Energy}$ | $m \rightarrow \text{mass}$ | $c \rightarrow \text{Speed of light} = 3 \times 10^8 \text{ velocity}$

~~$$mc^2$$~~

$$E = h\nu \quad | \quad E = mc^2$$

$$h\nu = mc^2$$

$$\frac{h \cdot \nu}{\lambda} = mc^2$$

$$\frac{h}{\lambda} = mc$$

$$\lambda = \frac{h}{mc} \quad \text{for } c$$

We can change c to v for normal particles.

$$\lambda = \frac{h}{mv}$$

$$\lambda = \frac{h}{p}$$

$$\Rightarrow \lambda = \frac{h}{\sqrt{2mE_k}} \quad p = mv \text{ (linear momentum)}$$

 m → mass of particle
 E_k → kinetic energy

$$\lambda = \frac{h}{\sqrt{2mqV}}$$

 q → charge of particle

 V → voltage

 m → mass of the particle

$$p = mv$$

$$p^2 = m^2 v^2$$

$$\frac{p^2}{2m} = \frac{m^2 v^2}{2m}$$

$$\frac{p^2}{2m} = KE$$

But $(KE = eV)$

$$\frac{p^2}{2m} = eV$$

$$p^2 = 2meV$$

$$p = \sqrt{2meV}$$

↑
for accelerated electron

$$e = +1.6 \times 10^{-19} \text{ C}$$

$$m = 9.1 \times 10^{-31} \text{ kg}$$

3b) $m = 9.1 \times 10^{-31} \text{ kg}$ $c = 3 \times 10^8 \text{ m s}^{-1}$ $KE = 100 \text{ eV}$ $m_{rel} = 0.5 \text{ meV}$
 $h = 6.6 \times 10^{-34} \text{ J s}$ $\frac{\text{C}^2}{\text{V}}$

$$\lambda = \frac{h}{mc}$$

$$m = \frac{0.5 \times 10^6 \times 1.6 \times 10^{-19} \text{ J kg}}{3 \times 10^8 \times 3 \times 10^8}$$

$$= \frac{0.8 \times 10^{-13}}{9 \times 10^{16}}$$

$$= 0.088 \times 10^{-29} \text{ kg}$$

$$\lambda = \frac{h}{\sqrt{2meV}}$$

$$\lambda^2 = (6.63 \times 10^{-34})^2$$

~~$$2 \times 9.1 \times 10^{-31} \times 2 \times 0.088 \times 10^{-29} \times 100 \times 1.6 \times 10^{-19}$$~~

$$\lambda^2 = 43.9569 \times 10^{-68}$$

$$28.16 \times 10^{-48}$$

$$\lambda^2 = 10^{-68+48} \times 1.560$$

$$\lambda^2 = 10^{-20} \times 1.560$$

$$\lambda^2 = (10^{-10})^2 \times (1.2489)^2$$

$$\lambda = (10^{-10} \times 1.2489)$$

$$\lambda = 1.2489 \times 10^{-10} \text{ m} / \therefore$$

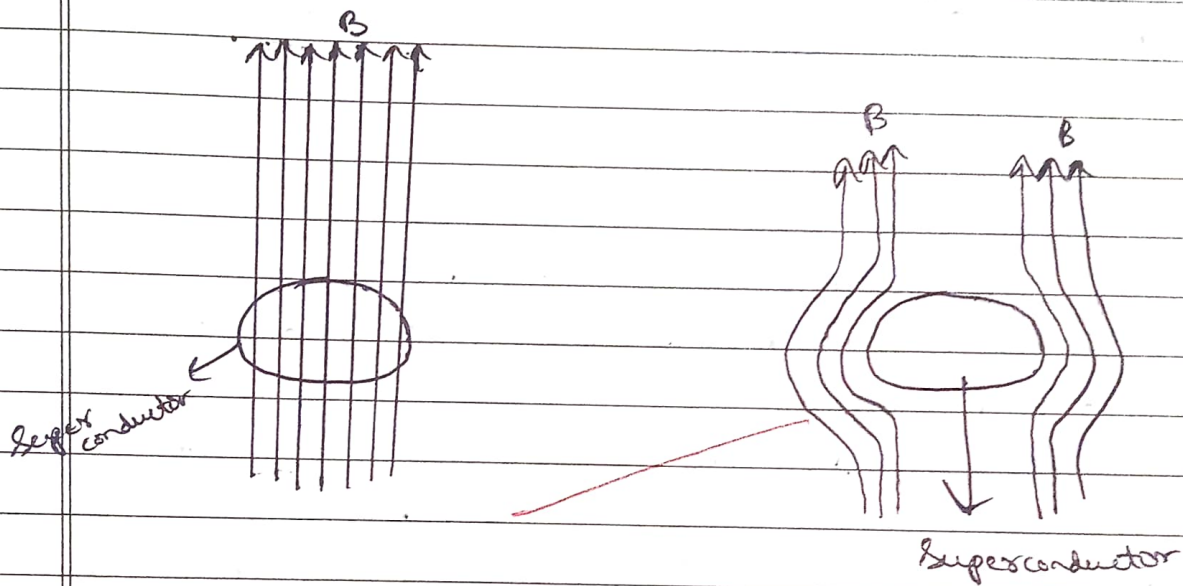
$$\lambda = 1.2489 \text{ \AA} / \therefore$$

2nd Internals

Part A

1A- When a superconductor is placed in a magnetic field at normal temperature (greater than critical temperature), the magnetic field lines pass through it.

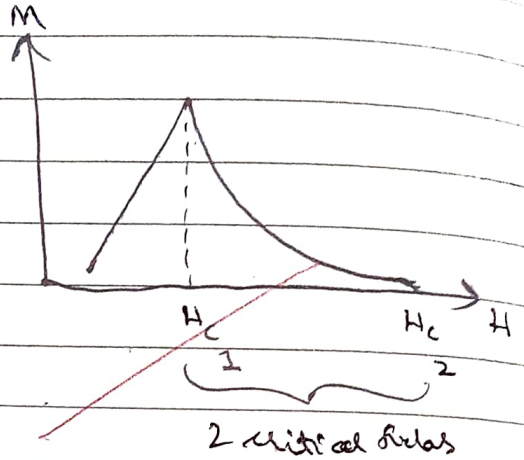
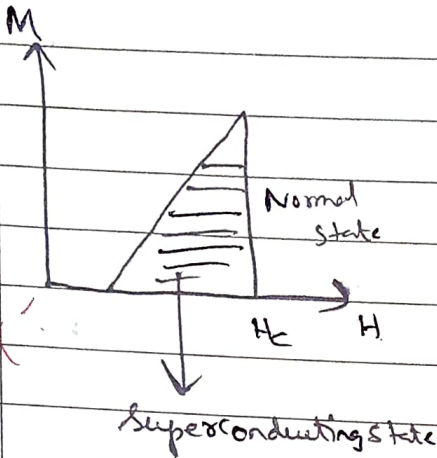
But when a superconductor is placed in the magnetic field when cooled below its critical temperature, the magnetic field lines get expelled out of it. This is called Meissner's effect.



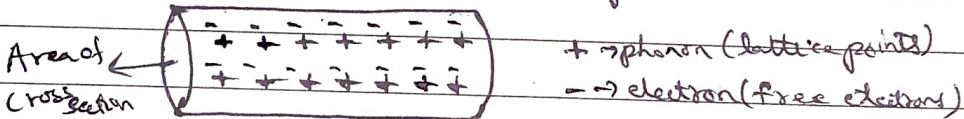
Superconductor Soft (Type-1)	Hard (Type-2) Superconductor
<ul style="list-style-type: none"> Exhibit Meissner's effect completely has 1 critical field No vortex region Low critical temperature Low critical magnetic field Can't be used for field magnets Eg: Zn, Al 	<ul style="list-style-type: none"> Exhibit Meissner's effect partially has 2 critical fields Vortex region present high critical temperature high critical magnetic field can be used for field magnets. Eg: Alloys of Niobium (Nb)

Type-1

Type-2



- b)
- It was proposed by Sommerfeld in 1928.
 - The energy levels of the electrons are quantized.
 - The allowed energy levels of the electrons are realized in terms of quantum states.
 - The order of energy levels is as per Pauli's exclusion principle.
 - The forces of attraction b/w the electrons of themselves and the force of ~~attr~~ repulsion b/w the scattered electron and the lattice points are ignored.
 - The ~~pe~~ electric potential inside the metal is assumed to be constant with the free electrons moving ^{with} elastic collisions continuously.



i) $T_c = 7.06 K$

$T = 6 K$

$H_0 = 50 \times 10^3 \text{ Am}^{-1}$

$H_c = ?$

$$H_c = H_0 \left[1 - \left(\frac{v}{T_c} \right)^2 \right]$$

$$H_c = 50 \times 10^3 \left[1 - \left(\frac{6}{7.26} \right)^2 \right]$$

$$H_c = 50 \times 10^3 \left[1 - \left(\frac{36}{52.7076} \right) \right]$$

$$04 \quad H_c = 50 \times 10^3 \left[1 - 0.68301 \right]$$

$$H_c = 50 \times 10^3 \left[0.31699 \right]$$

$$H_c = 15.8495 \times 10^3 \text{ Am}^{-1}$$

PART-B

3a) W.K.T

$$\lambda = \frac{h}{p} \rightarrow \text{① } \lambda \rightarrow \text{wavelength}$$

p

$$h \rightarrow \text{planck's constant} = 6.625 \times 10^{-34} \text{ Js}$$

p \rightarrow linear momentum mv

↓

This is de-Broglie's expression for wavelength.

$$\Psi = A e^{i(kx - \omega t)}$$

Exponential form of a
Total wave function

$$\Psi = A e^{ikx} \times e^{-i\omega t}$$

$$\Psi = \underbrace{\psi}_{\substack{\uparrow \\ \text{Space wave function} \\ \text{Only position coordinate} \\ \text{no time coordinate}}} \times e^{-i\omega t} \rightarrow \textcircled{3}$$

Ψ = Capital psi \rightarrow displacement
A \rightarrow Amplitude

e = log base

$$i = \sqrt{-1}$$

k \rightarrow wave number $= \frac{2\pi}{\lambda}$

x \rightarrow position coordinate

ω \rightarrow angular frequency

t \rightarrow time coordinate

Differentiating $\textcircled{3}$ w.r.t x twice, we get,

$$\frac{d\Psi}{dx} = e^{-i\omega t} \times \frac{d\psi}{dx}$$

$$\frac{d^2\Psi}{dx^2} = e^{-i\omega t} \times \frac{d^2\psi}{dx^2} \rightarrow \textcircled{4}$$

Differentiating $\textcircled{3}$ w.r.t t twice, we get,

$$\frac{d\Psi}{dt} = \psi \cdot e^{-i\omega t} \cdot (-i\omega)$$

$$\frac{d^2\Psi}{dt^2} = \psi \cdot e^{-i\omega t} \cdot (-i\omega) \cdot (-i\omega)$$

$$= i^2 \omega^2 \psi e^{-i\omega t}$$

$$= -\omega^2 \psi e^{-i\omega t} \rightarrow \textcircled{5}$$

$$i = \sqrt{-1}$$

$$i^2 = -1$$

For any travelling wave, with displacement ψ , the wave equation is given by,

$$\frac{d^2 \psi}{dx^2} = \frac{1}{v^2} \left(\frac{d^2 \psi}{dt^2} \right) \longrightarrow \textcircled{6}$$

This is the differential equation of the wave
 $y = A \sin(kx - \omega t)$

④ & ⑤ in ⑥, we get,

$$e^{-i\omega t} \frac{d^2 \psi}{dx^2} = \frac{1}{v^2} \left[-\omega^2 \psi e^{-i\omega t} \right]$$

$$\therefore \frac{d^2 \psi}{dx^2} = \frac{1}{v^2} (-\omega^2 \psi)$$

$v = f\lambda$ \downarrow velocity \uparrow frequency	\rightarrow wavelength	$\omega = 2\pi f$ \downarrow angular velocity \uparrow frequency
---	--------------------------	--

$$\therefore \frac{d^2 \psi}{dx^2} = \frac{1}{f^2 \lambda^2} (-4\pi^2 f^2 \psi)$$

$$\frac{d^2 \psi}{dx^2} = \frac{1}{\lambda^2} (-4\pi^2 \psi)$$

$$-4\pi^2 \psi \frac{d^2 \psi}{dx^2} = \frac{1}{\lambda^2} \longrightarrow \textcircled{7}$$

$$W.K.T \quad KE = \frac{p^2}{2m}$$

$$\left[p = \sqrt{2m \cdot KE} \quad p^2 = 2m \cdot KE \right]$$

But we also know that $p = \frac{h}{\lambda}$ (eq. ①)

$$KE = \frac{h^2}{\lambda^2 \cdot 2m}$$

~~$$KE = \frac{h^2}{2m} \left(\frac{1}{\lambda^2} \right)$$~~

$$KE = \frac{h^2}{2m} \left(\frac{-1}{4\pi^2} \frac{d^2\psi}{dx^2} \right)$$

$$KE = -\frac{h^2}{8\pi^2 m} \frac{d^2\psi}{dx^2} \rightarrow \text{②}$$

But Total Energy = Kinetic Energy + Potential Energy

$$E = KE + V$$

$$\therefore KE = E - V$$

$$E - V = -\frac{h^2}{8\pi^2 m} \frac{d^2\psi}{dx^2}$$

$$(E - V)\psi = -\frac{h^2}{8\pi^2 m} \frac{d^2\psi}{dx^2}$$

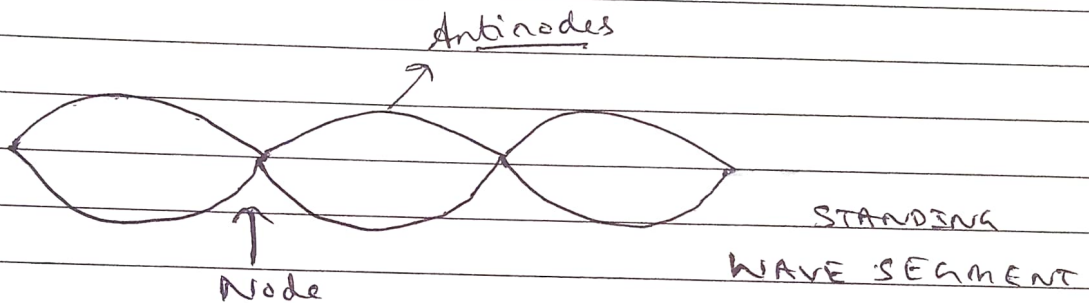
$$\frac{-8\pi^2 m \psi}{h^2} (E - V)\psi = \frac{d^2\psi}{dx^2}$$

$$\text{or } \frac{d^2\psi}{dx^2} + \frac{8\pi^2m}{h^2} (E - V)\psi = 0$$

↑
Time independent Schrodinger Wave equation

$$y(x, t) = A \cdot \sin\left(\frac{2\pi x}{\lambda}\right) \cdot \cos(2\pi ft)$$

↑
A standing wave can be expressed in this form.
This proves that electron is actually a standing wave around the nucleus



3b)

CLASSICAL COMPUTERS

- Data is stored in transistors in the form of bits in transistors
- Logic gates process data
- Power increases in the ratio of 1:1 with the increase in no. of transistors
- Can be used at room temperature
- Less errors
- Wide range of utility

QUANTUM COMPUTERS

- Data is stored as qubits (quantum bits). obeying all laws of quantum mechanics.
- quantum logic gates process data.
- Power exponentially increases with the increase in no. of qubits.
- can be used only at ultracold temperature
- more errors.
- used for data analysis, quantum

CLASSICAL COMPUTERS
to process calculations, data
Storage, critical problem
Solving etc.

QUANTUM COMPUTERS
computing and optimization problems

I have answered

1a)

3a)

1b)

3b)

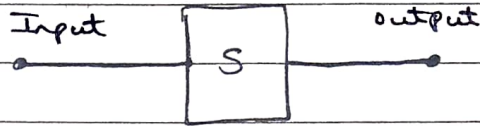
1c)

End -

Part - B

3a) Phase gate (S-gate)

• Block diagram:



• It is a single input gate

• $S = \begin{bmatrix} 1 & 0 \\ 0 & i \end{bmatrix}$

• Can be formed by connecting 2 T-gates in series.

• Calculation:-

$S|0\rangle = \begin{bmatrix} 1 & 0 \\ 0 & i \end{bmatrix} \times \begin{bmatrix} 1 \\ 0 \end{bmatrix} = \begin{bmatrix} 1 \\ 0 \end{bmatrix} = |0\rangle$

$S|1\rangle = \begin{bmatrix} 1 & 0 \\ 0 & i \end{bmatrix} \times \begin{bmatrix} 0 \\ 1 \end{bmatrix} = \begin{bmatrix} 0 \\ i \end{bmatrix} = i \begin{bmatrix} 0 \\ 1 \end{bmatrix} = i|1\rangle$

$S(\alpha|0\rangle + \beta|1\rangle) = \begin{bmatrix} 1 & 0 \\ 0 & i \end{bmatrix} \begin{bmatrix} \alpha \\ \beta \end{bmatrix} = \begin{bmatrix} \alpha \\ i\beta \end{bmatrix} = \alpha|0\rangle + i\beta|1\rangle$

Truth table:-

Input	Output
$ 0\rangle$	$ 0\rangle$
$ 1\rangle$	$i 1\rangle$
$\alpha 0\rangle + \beta 1\rangle$	$\alpha 0\rangle + i\beta 1\rangle$

It can be factored by connecting 2 gates in series

$$T = \begin{bmatrix} 1 & 0 \\ 0 & e^{i\pi/4} \end{bmatrix} = \begin{bmatrix} 1 & 0 \\ 0 & \frac{i+1}{\sqrt{2}} \end{bmatrix}$$

$$S = \begin{bmatrix} 1 & 0 \\ 0 & i \end{bmatrix}$$

$$T \times T = \begin{bmatrix} 1 & 0 \\ 0 & \frac{i+1}{\sqrt{2}} \end{bmatrix} \begin{bmatrix} 1 & 0 \\ 0 & \frac{i+1}{\sqrt{2}} \end{bmatrix}$$

$$T^2 = \begin{bmatrix} (1 \times 1) + (0 \times 0) & (1 \times 0) + 0 \times \left(\frac{i+1}{\sqrt{2}}\right) \\ (0 \times 1) + \left(\frac{i+1}{\sqrt{2}}\right) \times 0 & (0 \times 0) + \left(\frac{i+1}{\sqrt{2}}\right)^2 \end{bmatrix}$$

$$= \begin{bmatrix} 1 & 0 \\ 0 & \frac{(i+1)^2}{2} \end{bmatrix} = \begin{bmatrix} 1 & 0 \\ 0 & \frac{i^2 + 1 + 2i}{2} \end{bmatrix}$$

$$= \begin{bmatrix} 1 & 0 \\ 0 & 2i/2 \end{bmatrix} = \begin{bmatrix} 1 & 0 \\ 0 & i \end{bmatrix} = S //$$

$$\therefore S = T^2 //$$

$$3b) \quad |\psi\rangle = \begin{bmatrix} \alpha_1 \\ \alpha_2 \end{bmatrix} \quad |\phi\rangle = \begin{bmatrix} \beta_1 \\ \beta_2 \end{bmatrix}$$

$$\begin{aligned} \text{To prove: } \langle \psi | \phi \rangle &= \langle \phi | \psi \rangle^* \\ &= \langle \underbrace{\psi}_{\substack{\downarrow \\ \text{Bra} \\ \text{vector}}} | \underbrace{\phi}_{\substack{\downarrow \\ \text{ket} \\ \text{vector}}} \rangle = \left(\langle \phi | \psi \rangle \right)^* \end{aligned}$$

$$\langle \psi | = [\alpha_1, \alpha_2] \quad \langle \phi | = [\beta_1, \beta_2]$$

$$[\alpha_1, \alpha_2] \begin{bmatrix} \beta_1 \\ \beta_2 \end{bmatrix} = [\alpha_1 \beta_1 + \alpha_2 \beta_2] = \text{LHS}$$

04

1x2 2x1

$$[\beta_1, \beta_2] \times \begin{bmatrix} \alpha_1 \\ \alpha_2 \end{bmatrix} = [\beta_1 \alpha_1 + \beta_2 \alpha_2] = \text{RHS}$$

$$= [\alpha_1 \beta_1 + \alpha_2 \beta_2] = \text{RHS}/$$

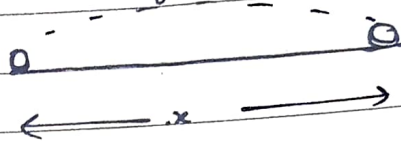
$x + iy \longrightarrow x - iy \longrightarrow$ only for imaginary part.
CONJUGATION

There is no i part in RHS. So, Conjugate of that remains same. $[\beta_1 \alpha_1 + \beta_2 \alpha_2]$ only/.

LHS = RHS/.

Part-A

1a) Motion ^{of a body} along a straight line is called linear motion after time t



Uniform motion is the motion of a body where equal distances are covered in equal intervals of time.

Slow in implies acceleration [rate of change of increase in velocity w.r.t time]

Slow out implies deceleration [rate of change of decrease in velocity w.r.t time]

03 ✓
 Speed = $\frac{\text{distance}}{\text{time}}$ velocity = $\frac{\text{displacement}}{\text{time}}$

$$s = \frac{d}{t}$$

$$v = \frac{x}{t}$$

$$v = \frac{\Delta x}{\Delta t}$$

$$v = \lim_{\Delta t \rightarrow 0} \frac{\Delta x}{\Delta t} = \frac{dx}{dt}$$

$$\text{acceleration} = \frac{\text{velocity}}{\text{time}} = \frac{v}{t} = \frac{\Delta v}{\Delta t} = \lim_{\Delta t \rightarrow 0} \frac{\Delta v}{\Delta t} = \frac{dv}{dt}$$

3 kinematic equations:-

$$v = u + at$$

$$s = ut + \frac{1}{2} at^2$$

$$v^2 - u^2 = 2as$$

1c) Total distance = (last frame no - 1) × base distance

$$25 \text{ m} = (6 - 1) \times x$$

$$25 = 5 \times x$$

$$\therefore x = 5$$

$$\therefore \text{Base distance} = 5 \text{ m}.$$

1b) Push time is the time taken to push a body upto a certain distance. It involves acceleration.

Stop time is the time taken to stop or obstruct the motion of the body. It involves retardation.

The abrupt stop or halt of a body ^{against} on a surface results in collision.

$$v_2 - v_1 = e(u_1 - u_2)$$

↑ final velocity
 ↑ elasticity

Newton's Law of Collision.

3 Types of Collision: Perfectly elastic
Inelastic
Perfectly inelastic

~~Conservation~~ Conservation of linear momentum plays a vital role in collision.

$$m_1 u_1 + m_2 u_2 = m_1 v_1 + m_2 v_2$$

End